

Second Year of M. Sc.(Physics)  
M. Sc. II, Semester – IV

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MC 202: ORGANIC CHEMISTRY – I	3	1	2	5

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- **CYCLOALKANES AND DIENES** (04 Hours)  
Cycloalkanes: Nomenclature, methods of formation, chemical reactions, Baeyer's strain theory and its limitations, theory of strainless ring. Dienes: Nomenclature, classification, methods of formation of butadiene, chemical reactions, 1,2 and 1,4 additions, Diel's – Alder reaction.
  - **BENZENE AND ITS HOMOLOGUES** (06 Hours)  
Aromaticity, Mobius and Huckel polyenes, Huckel rule, annulene, mechanism of substitution reactions, directive effects of substituents, *o*, *p* and *m*-directing groups, effect of substituents on reactivity, theory of activity and deactivity effects. Fused ring compounds: Chemistry of naphthalene, anthracene and phenanthrene.
  - **UNIT PROCESSES** (06 Hours)  
Sulphonation: Definition, methods, sulphonating agents, sulphonation of benzene with the help of SO<sub>3</sub>, H<sub>2</sub>SO<sub>4</sub>, oleum and ClSO<sub>3</sub>H Nitration: Definition, methods, nitrating agents, factors affecting nitration, nitration of benzene, naphthalene, importance of nitration in manufacture of artificial perfume.
  - **HETEROCYCLIC COMPOUNDS** (07 Hours)  
Nomenclature, aromaticity, synthesis, properties, uses and canonical structures of pyrrole, furan and thiophene.
  - **CARBOHYDRATES** (05 Hours)  
Introduction to disaccharides, glycosidic bond, structure determination of sucrose, lactose, maltose and cellobiose.
  - **PHOTOCHEMISTRY** (06 Hours)  
Laws of photochemistry, nature of electronically excited states, geometry, dipole moment, acid base properties, internal conversion, intersystem crossing, phosphorescence, fluorescence, quantum yield, examples of low and high quantum yield, actinometry, rate of photochemical reactions, photochemical reactions of >C=C<, >C=O, benzene ring, and nitrogen containing compounds, photooxygenations, photochemistry of air and air pollution, chemi- and bioluminescence.
  - **COMPOUND CONTAINING ACTIVE METHYLENE GROUP** (02 Hours)  
Malonic ester and acetoacetic ester: preparation and its synthetic applications.
  - **PETROCHEMICALS** (06 Hours)  
Petrochemicals obtained from C<sub>1</sub> cut of petroleum, manufacture and applications of ammonia, formaldehyde, hexamethylene tetramine, chlorinated methane. Petrochemicals obtained from C<sub>2</sub> cut of petroleum, manufacture and applications of chemicals obtained from ethanol, acetaldehyde (Wacker Cheime Process), ethylene dioxide, ethylene glycol.

(Total Contact Time (Theory): 42 Hours)

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**BOOKS RECOMMENDED:**

1. Morrison R. T. and Boyd R.N., 'Organic Chemistry', 6th Edn., Prentice Hall, 1992.
2. Bahl A. and Bahl B. S., 'A Textbook of organic Chemistry', 2<sup>nd</sup> Edn., S. Chand, 2005.
3. Kumar S., 'Introduction to Petrochemicals'. 6<sup>th</sup> Edn, Oxford & IBH, 2000.
4. March J., 'Inorganic Chemistry', 5<sup>th</sup> Edn., S. Chand, 2001.
5. Finar I. L., 'Organic Chemistry' volume 1 & 2 6<sup>th</sup> edition Longman, London 2006.