First year of Five Years integrated M. Sc. (Physics / Chemistry / Mathematics)

M.Sc. - I, Semester – I L T P C

MC 101: Chemistry - I

ATOMIC STRUCTURE AND CHEMICAL BONDING

(13 Hours)

5

2

1

3

Heisenberg's Uncertainity principle, postulates of quantum mechanics, Schrödinger wave equation: Derivation, significance of and ψ^2 , Schrödinger wave equation for H - atom and particle in 1-D box, angular and radical wave function, atomic orbitals and shape of s and p orbitals, Valence bond theory, Hybridization, Resonance, VSEPR, Molecular orbital theory, molecular orbitals, bonding and energy level diagram for homonuclear and heteronuclear diatomic molecules, ionic solids, Born-Haber cycle, covalent bonds, coordinate bond, hydrogen bond, dipole moment, geometry and shape of simple molecules, Molecular structure and different kind of intermolecular forces and interactions like hydrogen bonding, hydrophobicity, π - π interaction, π -cation interaction, and properties such as melting and boiling points, dipole moment, acidity and basicity.

PERIODIC TABLE AND ATOMIC PROPERTIES

(03 Hours)

Electronic configuration, periodicity in properties: ionization potential, electron affinity, ionic radii and electronegativity

CHEMICAL KINETICS

(04 Hours)

Rate of reaction, order of reaction, enzyme catalyzed reaction, fast reactions, homogeneous and heterogeneous catalysis, general characteristics of catalytic reactions.

THERMODYNAMICS

(07 Hours)

First law of thermodynamics, entropy, second and third laws of thermodynamics, Gibbs free energy, Helmholtz energy, chemical equilibria, Clausius Clapeyron equation

ELECTROCHEMISTRY

(08 Hours)

Single electrode potential, Hydrogen electrode, Galvanic cell, EMF series, Nernst equation, Reversible electrodes, metal-metal ion electrodes, Calomel electrode, Oxidation-Reduction electrodes, Potentiometric titration, Application of electrochemistry in Corrosion control by cathodic protection, batteries, and related devices.

CHEMISTRY OF WATER

(07 Hours)

Structure of water, properties, types of water (raw water, cooling water, boiler water), role of water in life sciences, Water-treatment- primary treatment and secondary treatment, types of water treatment for use in industries.

(Total Contact Time (Theory): 42 Hours)

BOOKS RECOMMENDED:

- 1. **Atkins P. W. and Paula D**., *Atkin's Physical Chemistry*, Oxford University Press/Gopsons Paper Ltd, Noida, 8th Edn., 2006.
- 2. Alberty R. A. and Silbey R. J., *Physical Chemistry*, 1st Edn., John Wiley & Sons (Asia), Singapore, 1995.
- 3. Levine I. R., Quantum Chemistry, Prentice Hall India (Ltd), 1995.
- 4. Lee J. D., Concise Inorganic Chemistry, 4th Edn., ELBS, 1991.
- 5. Cotton F. A., Wilkinson G., Gans P. G., Basic Inorganic Chemistry, 2nd Edn., John Wiley & Sons, 1987.